

## FRAUNHOFER INSTITUTE FOR CHEMICAL TECHNOLOGY ICT



 Evaluation of safe and highperformant bromine cathodes: hydrogen-bromine single cell.
Principle of hydrogen-bromine redox flow battery.

3 Safe energy storage through electrolyte development: Storage of charged bromine with bromine complexation agent in the form of a two-phase liquid system vs. pure HBr/Br<sub>2</sub> electrolyte.

## Fraunhofer Institute for Chemical Technology ICT

Joseph-von-Fraunhofer-Strasse 7 76327 Pfinztal Germany

Contact

Michael Kuettinger Phone +49 721 4640-518 michael.kuettinger@ict.fraunhofer.de

www.ict.fraunhofer.de





## **BROMINE REDOX FLOW BATTERIES** RESEARCH AND APPLICATION

Redox flow batteries are finding increasing application in energy storage. Bromine-based redox flow battery systems are regarded as one of the nextgeneration redox flow battery technologies. With our expertise we provide our customers with research and development services for bromine-based redox flow batteries, from basic and applied research up to system solutions.

We assist our customers in the development of bromine-based redox flow batteries. In our research we meet safety requirements for bromine use through a combination of elaborated chemistry and engineering. In order to improve the battery performance we are focusing on both converter cell and storage systems.

## Our offer

- Development of bromine-based
- electrolyte formulation.
- R&D of half cells and single cells in different bromine-based chemistries – H<sub>2</sub>/Br<sub>2</sub>, Zn/Br<sub>2</sub> and other customized cell chemistry. Process engineering on gaseous/liquid(/liquid) fluidic system to improve charge and discharge performance.

- Testing and developing customized cell components for bromine-based batteries.
- Developing analytical methods and diagnostic tools for quantification according to customer requirements.
- Drafting and execution of feasability studies for different bromine cell chemistries and recommendations for product development.
- Flexibile work on customer-specific questions.
- Methods: Half-cell electrochemistry cell performance, including in-situ half-cell performance with various electrochemical methods; surface survey and electrolyte survey using analytics and spectroscopic methods.